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Lipophilic character of cardiac glycosides

R_M values as lipophilicity parameters

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SUMMARY

Chromatographic R_M values were measured for a series of 41 cardiac glycosides and aglycones. By means of the AR_M values it was possible to calculate the R_M values for a further 19 compounds. An excellent correlation was found between the present *RM* values and those measured or calculated with the Cohnen et al. system. In a similar way, the R_M values were shown to be well correlated with both high-performance liquid chromatographic data and octanol-water partition coefficients (log P). The additive contribution of each substituent group to the overall lipophilicity of the molecule seems to be constant in each subset of *Digitalis* derivatives.

INTRODUCTION

The cardiac steroids, despite their long history, are among the most important drugs in modern therapeutics. However, quantitative studies of the relationships between structure and activity (QSAR) of these compounds have received little attention¹⁻³. Only in recent times has there been some increasing interest in $OSAR$ studies of cardiac steroids^{$4-9$}. As the lipophilic character is one of the most important physico-chemical factor determining the biological activity of drugs, several workers^{4,6,7,10} have studied the determination of lipophilicity indices of cardiac steroids. Nevertheless, there is still a lack of data describing the lipophilic character of most of these compounds. The determination of the classical octanol-water partition coefficient is the main method for establishing the lipophilic character of drugs. However, the R_M and log k' values obtained by reversed-phase partition thin-layer (TLC) and high-performance liquid chromatography (HPLC) were shown to be well correlated with the partition coefficients of many chemicals and have been proposed as an alternative method^{11,12}. The R_M values were calculated by means of the equation

$$
R_M = \log\left(\frac{1}{R_F} - 1\right)
$$

An earlier contribution to the study of the relationship between chemical structure and chromatographic behaviour was provided by Nover and co-wor $kers^{13,14}$ by means of adsorption paper and thin-layer chromatography. With regard to a QSAR study, the purpose of this work was to determine or calculate R_M and log k' values for a large series of cardiac steroids.

EXPERIMENTAL

Chemicals

Several cardiac glycosides and aglycones (compounds 4,8,41,43,44,45,46,55, 57, 58 and 60) were a generous gift from Simes (Milan, Italy). Other compounds were obtained from commercial sources. All the drugs were used without further purification. All solvents were of analytical-reagent or HPLC grade.

Determination of RM values

The TLC technique had been described previously¹⁵. Glass plates (20 \times 20 cm) were coated with silica gel G (E. Merck, Darmstadt, F.R.G.). In order to control the pH of the stationary phase, a slurry of silica gel G was prepared with 0.09 M sodium hydroxide solution. A non-polar stationary phase was obtained by impregnating the silica gel G layer with silicone DC 200 (viscosity 350 cSt) (Applied Science Labs., State College, PA, U.S.A.). The impregnation was carried out by developing the plates in a 5% silicone solution in diethyl ether. Eight plates could be impregnated in a single chromatographic chamber containing 200 ml of the silicone solution. The plates were left in the chamber for 12 h, *i.e.,* for several hours after the silicone solution had reached the top of the plates. The chromatographic chamber was saturated with the vapour of the mobile phase.

A migration distance of 10 cm was obtained on all plates by cutting the layer at 12 cm and spotting the compounds on a line 2 cm from the lower edge of the plate. The mobile phase saturated with silicone oil was aqueous buffer (sodium acetate-Verona1 buffer, $1/7 M$ at pH 7.2), alone or mixed with various amounts of acetone. Two plates were developed simultaneously in a chromatographic chamber containing 200 ml of mobile phase, at room temperature.

The cardioactive drugs were dissolved in methanol, acetone or water $(1-2)$ mg/ml) and 1 μ of solution was spotted randomly on the plates in order to avoid any systematic error. The developed plates were dried and sprayed with an alkaline solution of potassium permanganate. After a few minutes at 12o"C, yellow spots appeared on an intense pink background.

Determination of log P values

The n-octanol-water partition coefficients of six genins were measured by means of the shake-flask method¹⁶, using distilled water as the polar phase and *n*-octanol as the lipid phase; each phase was previously saturated with the other one and centrifuged if not clear.

A carefully weighed amount of compound was dissolved in the octanol phase, and an appropriate amount of water was added; the bottles were then shaken gently for ca. 5 min and centrifuged for 1.0 h at 2000 rpm $(895 g)$. The ratio of the octanol and water volumes was chosen so as to give a reliable UV absorbance at the wavelength of maximum absorption.

The concentration in the aqueous phase was determined spectrophotometrically by means of a Varian DMS-90 UV-visible spectrophotometer; each reported log *P* value is the average of at least four determinations with $s \le 0.03$.

Determination of log k' values

HPLC was performed on a Waters Assoc. 820 chromatography workstation, using a µBondapak C₁₈ column (300 \times 3.9 mm I.D.) (Waters Assoc., Milford, MA, U.S.A.), packed with silica gel (particle size 10 μ m) with a C₁₈ chemically bonded non-polar stationary phase¹⁷. A Waters Assoc. UV detector at 220 nm and Hamilton 802 chromatographic syringes (25 μ) were used. The compounds were dissolved in methanol (1 mg/ml) and applied to the column in 5- μ l volumes. All solutions and reagents were first filtered through Millipore filters (Type FH, pore size $0.5 \mu m$). The separation was carried out using acetonitrile-water mixtures as the mobile phase at a flow-rate of 1 ml/min. The acetonitrile concentration ranged from $20-40$ to $50-70\%$. The retention times were expressed as

$$
\log k' = \log \left(\frac{t_x - t_0}{t_0} \right)
$$

where t_x is the retention time of the compound and t_0 that of the solvent front.

RESULTS

RM values and lipophilicity of cardioactive steroids

The reversed-phase TLC of the cardioactive compounds showed that most of them did not move from the starting line when the mobile phase was aqueous buffer alone. In order to obtain suitable R_M values it was necessary to add acetone to the mobile phase. Only with the six most hydrophilic compounds, ouabain, strophanthidol, k-strophanthidin, k-strophanthin β , k-strophanthoside and digoxigenin, could reliable R_M values be obtained even at 0% acetone in the mobile phase. However, as usual in TLC and HPLC, for all the compounds there was a linear relationship between *RM* values and a range of acetone concentration.

The equations describing such a linear relationship allowed the calculation of extrapolated R_M values at 0% acetone in the mobile phase for the compounds that did not migrate with the aqueous buffer alone. The validity of the extrapolation technique is shown by the fact that the experimental R_M values of 0% acetone of the above six most hydrophilic compounds are very close to the extrapolated *RM* values, calculated for the same compounds over a wider range of acetone concentrations. All the extrapolated R_M values are reported in Tables I and II.

The range of the linear relationship between R_M values and acetone concen-

 R_M VALUES OF CARDIAC GLYCOSIDES AND AGLYCONES OF DIGITALIS SPP.

TABLE I

66

G. L. BIAGI et al.

y Calculated *Ru* values (see text)

^b Evomonoside, neriifolin, oleandrigenin, oleandrin and 16-desacetyloleandrin are listed in Table I as they are characterized by Digitalis aglycones. b Evomonoside, neriifolin, oledndrigenin, oleandrin and 16-desacetyloleandrin are listed in Table I as they are characterized by *Digitalis* aglycones.

e TLC system: see text. ' TLC system: see text

168

TABLE II

" Calculated R_M values (see text).
" TLC system: see text.

trations is limited by the fact that at lower and higher acetone concentrations all the compounds tend not to move from the starting line or to migrate with the solvent front, respectively, *i.e.*, to deviate from the linear relationship. Therefore, the extrapolated R_M values in Tables I and II were obtained from equations calculated by means of R_M

TABLE III

 \mathbf{t}

RANGES OF ACETONE CONCENTRATIONS USED FOR THE CALCULATION OF THE TLC EQUATIONS

Compound No.	Acetone range $(%$	TLC equation		
		$a = R_M$	b	r
57	$0 - 24$	0.535 ± 0.019	-0.066 ± 0.001	0.996
51	$0 - 24$	0.732 ± 0.071	-0.033 ± 0.002	0.985
41	$0 - 24$	0.938 ± 0.098	-0.038 ± 0.038	0.967
$\mathbf{1}$	$0 - 32$	0.980 ± 0.047	-0.048 ± 0.002	0.992
45	$0 - 32$	0.991 ± 0.162	-0.058 ± 0.007	0.961
44	$0 - 36$	$1.022 + 0.184$	-0.063 ± 0.011	0.959
52	$4 - 32$	1.194 \pm 0.043	-0.050 ± 0.002	0.995
49	4-36	1.213 ± 0.080	-0.045 ± 0.003	0.982
42	4-40	1.230 ± 0.100	-0.043 ± 0.003	0.987
$\overline{2}$	$4 - 24$	1.240 \pm 0.118	-0.063 ± 0.007	0.972
46	$4 - 32$	$1.295 + 0.090$	$-0.059 + 0.004$	0.983
43	$4 - 24$	1.343 ± 0.085	-0.065 ± 0.005	0.986
38	$8 - 36$	1.581 ± 0.070	-0.043 ± 0.002	0.991
3	$8 - 32$	1.650 ± 0.112	-0.063 ± 0.005	0.983
60	$8 - 50$	1.684 ± 0.117	-0.050 ± 0.004	0.976
26	$8 - 50$	1.719 ± 0.044	-0.050 ± 0.001	0.992
8	$8 - 40$	1.859 ± 0.131	-0.068 ± 0.005	0.981
15	$16 - 50$	$1.932 + 0.058$	$-0.050 + 0.002$	0.996
$\overline{\mathbf{4}}$	$16 - 50$	2.010 ± 0.094	$-0.062\,\pm\,0.003$	0.989
25	$16 - 50$	2.141 ± 0.221	-0.055 ± 0.006	0.986
58	$16 - 50$	2.149 \pm 0.120	-0.054 ± 0.003	0.985
7	$16 - 50$	2.187 ± 0.120	-0.066 ± 0.003	0.990
16	$20 - 50$	$2.232 + 0.122$	-0.058 ± 0.004	0.990
5	$20 - 50$	2.268 ± 0.118	-0.063 ± 0.003	0.994
39	$20 - 50$	2.345 ± 0.139	-0.057 ± 0.004	0.989
6	$24 - 50$	2.377 ± 0.119	-0.067 ± 0.003	0.994
13	$20 - 50$	$2.477 + 0.057$	-0.070 ± 0.002	0.985
12	$20 - 50$	2.503 ± 0.095	-0.071 ± 0.003	0.995
37	20–60	2.504 ± 0.124	$-0.062 \,\pm\, 0.003$	0.991
17	$28 - 55$	2.653 ± 0.147	-0.065 ± 0.003	0.993
9	$28 - 55$	2.887 \pm 0.122	-0.070 ± 0.003	0.994
36	$28 - 50$	2.933 ± 0.285	-0.068 ± 0.009	0.976
10	$28 - 55$	2.962 ± 0.209	$-0.072~\pm~0.005$	0.988
30	$32 - 40$	3.000 \pm 0.016	-0.076 ± 0.001	0.999
33	$28 - 50$	3.121 \pm 0.113	-0.078 ± 0.003	0.996
18	$28 - 60$	$3.183 + 0.141$	-0.070 ± 0.003	0.992
21	$28 - 55$	3.380 \pm 0.120	-0.072 ± 0.003	0.996
55	$28 - 50$	3.406 \pm 0.120	-0.066 ± 0.003	0.996
11	$28 - 50$	3.446 ± 0.303	-0.081 ± 0.008	0.985
19	$36 - 50$	3.540 \pm 0.450	-0.074 ± 0.010	0.981
35	$40 - 60$	4.588 ± 0.800	-0.088 ± 0.017	0.966

values determined with acetone concentrations ranging from 0 to 24% or from 36-40 to 50-60% depending on the lipophilicity of the test compounds. The most hydrophilic compound, ouabain, shows a linear relationship between R_M values and acetone concentration in the range O-24%. For the most lipophilic compound, pentaacetylgitoxin, acetone concentrations ranging from 40 to 60% were used.

The ranges of acetone concentrations and the TLC equations are reported in Table III, where a and b are the intercept and slope, with their standard errors, respectively, and r is the correlation coefficient. The intercepts $(a=R_M)$ are also reported in Tables I and II. In Table III the compounds are listed in order of increasing lipophilicity to show the good correlation between extrapolated R_M values and ranges of acetone concentrations. The slopes in Table III show that the equations describe a series of almost parallel straight lines.

Cohnen *et al.*¹⁰ measured the R_M values of a series of cardioactive steroids by means of a TLC technique which seems to be very similar to our own system. Their extrapolated R_M values at 0% acetone in the mobile phase are reported in Tables I and II. A very good correlation is shown by eqn. 1 between the present R_M values and those obtained by Cohnen *et al. lo* for a series of 23 compounds for which the experimental R_M values were available in both TLC systems.

$$
R_M = 0.361 \left(\pm 0.063 \right) + 0.937 \left(\pm 0.033 \right) R_{M\text{Cohnen}} (n = 23; r = 0.987; s = 0.121; F = 784.7; P < 0.005
$$
) (1)

In eqn. 1 and all subsequent equations, n is number of data points, r is the correlation coefficient, s is the standard error of the equation and *F* is the value of the F-test.

The experimental R_M values provide some understanding of the influence of substituent groups determining the lipophilicity of the whole molecule. The *Digitalis* glycosides can be grouped into three families on the basis of the aglycones (Table I). Cardiac glycosides and aglycones from different botanical sources are listed in Table II. The presence of digitoxosyl, acetyl or methyl group(s) in the sugar residue at C-3 increases the lipophilic character. The AR_M values reported in Table IV were used in the calculation of the R_M values of the mono- and bisdigitoxosides and also acetyl derivatives of trisdigitoxosides for which the experimental *RM* values were not available. In the Cohnen *et al.* system the R_M value of helveticoside was obtained by adding the average AR_M value of the digitoxosyl group to the R_M value of strophanthidin.

The R_M values of desacetyllanatoside C, strophanthidin-3-acetate, pentaacetylgitoxin and heptaacetyl-k-strophanthoside in the Cohnen *et al.* system were calculated from the experimental R_M values for lanatoside C, strophanthidin, 16-acetylgitoxin and k-strophanthoside by subtracting or adding an average AR_M of 0.29 for each of the acetyl groups.

$$
R_M \text{ (lanatoside B)} = R_M \text{ (gitoxin)} + [R_M \text{ (lanatoside A)} - R_M \text{ (digitoxin)}]
$$

= 2.83 + (3.09 - 2.92) = 3.00

$$
R_M \text{ (lanatoside B)} = R_M \text{ (gitoxin)} + [R_M \text{ (lanatoside C)} - R_M \text{ (digoxin)}]
$$

= 2.83 + (1.84 - 1.78) = 2.89

$$
\bar{x} = 2.94
$$

TABLE IV

INFLUENCE OF SUBSTITUENT GROUPS ON THE LIPOPHILIC CHARACTER

(Continued on p. I72)

 $2.11 - 2.1$

The AR_M value of the glucosyl group in the sugar residue was used for the calculation of the R_M values of glucogitoroside, erysimoside, scillaren A, k-strophanthin β and k-strophanthoside by adding it for one or two residues to the R_M values of gitoxigenin monodigitoxoside, helveticoside, proscillaridin and cymarin. The R_M values of evomonoside and ouabagenin were calculated from the R_M values of digitoxigenin and ouabain by adding or subtracting, respectively, the *RM* value of the rhamnosyl group in the sugar residue.

The AR_M value of the acetoxyl group vs. OH at C-16 was used in the calculation of the *RM* values of 16-acetylgitoxin and desacetyloleandrin by adding or subtracting it, respectively, from the R_M values of gitoxin and oleandrin.

The R_M values of helveticosol, convallatoxol, strophanthidol and convallatoxin were calculated by adding or subtracting the AR_M value for CH_2OH vs. CHO at C-10 from the R_M values of helveticoside, convallatoxin, strophanthidin and convallatoxol.

The R_M values of desglucocheirotoxin and olitoriside were calculated in both systems by adding the AR_M values of gulomethylose, or boivinose and glucose, respectively. As gulomethylose and boivinose are the isomeric forms of rhamnose and digitoxose, respectively, the AR_M values of the latter forms were used (Table IV). In fact, Davydov $\bar{6}$ had calculated the same retention values for both pairs of isomers. However, the results for another pair of isomers must be pointed out, i.e., cymarose and oleandrose, whereas $Davydov⁶$ had calculated the same retention values for these sugar residues, in the present and the Cohnen et al. system the AR_M values are different.

Finally, the R_M values of α -methyldigoxin, β -methyldigitoxin and α, β -dimethyldigoxin were calculated by means of the AR_M values for the α - and/or β -methyl groups in the side-chain.

The equation

$$
R_M = 0.202 \left(\pm 0.046 \right) + 1.027 \left(\pm 0.021 \right) R_M \text{ Cohen}
$$

(*n*=56; *r*=0.988; *s*=0.153; *F*=2265.3; *P* <0.005) (2)

calculated by means of both experimental and calculated R_M values is very similar to eqn. 1 and shows that the calculated R_M values do not deviate from the relationship described by eqn. 1 for the experimental R_M values. The slopes in both eqns. 1 and 2 are very close to unity, which explains why the AR_M values in Table IV are fairly close in the two systems.

Hence any substituent group tends to induce the same variation of lipophilic character in both TLC systems. Intercepts higher than zero indicate a systematic difference between the two systems, probably due to the different kind of silicone oil used by Cohnen *et al.* ¹⁰. Eqn. 2 holds over a wide range of R_M values, with a difference on a logarithmic scale of 4.65, which means a 44 668-fold difference in lipophilicity.

More recently, Dzimiri *et al.*⁷ measured the R_M values of a series of cardiotonic steroids by means of a different TLC system, which had also been used by Cohnen et al.¹⁰. The stationary phase was characterized by the presence of octanol instead of silicone oil. The mobile phase was methanol-water (30:70), which yielded the experimental R_M values reported in Tables I and II. When a compound had been tested also by Cohnen *et al.*¹⁰, an average R_M value is reported in Tables I and II.

The following equation describes the relationship between our R_M values and those measured by Dzimiri *et al.*⁷:

$$
R_M = 1.477 \left(\pm 0.092 \right) + 1.331 \left(\pm 0.105 \right) R_M \text{ Dzimiri}
$$
\n
$$
(n = 28; r = 0.927; s = 0.365; F = 159.0; P < 0.005)
$$
\n
$$
(3)
$$

The correlation coefficient is not as good as that for eqns. 1 and 2. A better equation was obtained when compounds 1, 15, 16, 30 and 35, which showed the largest deviations, were excluded from the analysis:

$$
R_M = 1.532 \left(\pm 0.064 \right) + 1.286 \left(\pm 0.078 \right) R_M \text{ Dzimiri}
$$
\n
$$
(n = 23; r = 0.963; s = 0.240; F = 269.8; P < 0.005)
$$
\n
$$
(4)
$$

A justification for excluding those compounds might be that they were the most deviant also when correlating the present R_M values with Dzimiri *et al.*'s log P values

OCTANOL-WATER PARTITION COEFFICIENTS AND HPLC RETENTION DATA FOR CARDIAC GLYCOSIDES

(eqn. 5). In the correlation with the $\log k'$ values (see eqn. 8) they were not excluded, but again they were among the most deviant compounds.

The higher intercepts in eqns. 3 and 4 are due to the fact that Dzimiri et al.⁷ used a stationary phase containing octanol instead of silicone oil. The higher slopes are due to the narrower range of Dzimiri *et al's R_M* values. In eqns. 3 and 4 about 7–14% of the variance in our R_M values is not explained by the regression.

Relationship between RM and log P or HPLC data

The log *P* and log *k'* values of the cardiac glycosides are reported in Table V, where most of the data available in the literature are also listed. Cohnen *et al.*¹⁰ and Dzimiri *et al.*⁷ measured the octanol-water partition coefficients of cardiac steroids. The best correlation between R_M and log \overline{P} values was found by Dzimiri *et al.*⁷. Therefore, in Table V we report only the log *P* values of the compounds for which an *R_M* value in Dzimiri *et al.*'s system was available. For the compounds tested by both Cohnen *et al.*¹⁰ and Dzimiri *et al.*⁷ an average log *P* is reported in Table V. Eqn. 5, excluding compounds 1, 15, 16, 30 and 35, and eqn. 6, considering only the six genins, can be compared with eqn. 7, calculated with Dzimiri *et al.*'s R_M and log *P* values:

$$
R_M = 1.248 \left(\pm 0.118 \right) + 0.634 \left(\pm 0.061 \right) \log P_{\text{Dzimiri}} \tag{5}
$$

(*n*=23; *r*=0.914; *s*=0.362; *F*=107.0; *P*<0.005)

$$
R_M = 0.345 \left(\pm 0.145 \right) + 0.637 \left(\pm 0.088 \right) \log P
$$

(n=6; r=0.964; s=0.193; F=52.2; P<0.005) (6)

$$
R_{M \text{ Dzimir}} = -0.252 \left(\pm 0.068 \right) + 0.516 \left(\pm 0.034 \right) \log P_{\text{Dzimir}} \tag{7}
$$
\n
$$
(n = 28; r = 0.948; s = 0.215; F = 233.5; P < 0.005)
$$

The different intercepts in eqns. 5 and 6 compared with eqn. 7 are due to the use of octanol instead of silicone oil in the Dzimiri et *al.* TLC stationary phase. The difference between the intercepts in eqns. 5 and 6 is due simply to the high standard error in eqn. 5 and to the fact that the two equations share only two compounds (nos. 26 and 41). On the other hand, the slopes of the three equations are very close.

Obviously the results with eqns. 5 and 7 could have been expected on the basis of eqn. 4. The log k' values reported in Table V were extrapolated to 0% acetonitrile in the mobile phase from the linear relationship between $\log k'$ and acetonitrile concentration as already described¹⁷. In Table V the HPLC data measured by Davydov⁶ and Dzimiri *et al.*⁷ are also listed. The relationship between the R_M values and the HPLC retention data is described by the following equations:

$$
R_M = -1.488 \left(\pm 0.192 \right) + 2.109 \left(\pm 0.110 \right) \log k'
$$
\n
$$
(n = 44; r = 0.947; s = 0.306; F = 369.5; P < 0.005)
$$
\n
$$
(8)
$$

$$
R_M = -0.249 \left(\pm 0.170\right) + 0.716 \left(\pm 0.054\right) \ln V
$$

(n=21; r=0.950; s=0.297; F=176.1; P<0.005) (9)

$$
R_M = 2.109 \left(\pm 0.064 \right) + 1.680 \left(\pm 0.121 \right) \log k'_{\text{Dzimiri}} (n=28; r=0.938; s=0.336; F=192.0; P<0.005)
$$
\n(10)

The $\ln V$ term in eqn. 9 is the retention index used by Davydov⁶.

In eqns. 5 and 6 and 8–10, again about $7-17\%$ of the variance in the R_M values is not explained by the regression. Nevertheless, the log *P* values and the HPLC data from two and three different laboratories, respectively, seem to agree with the R_M values as lipophilic indices of the cardiac steroids.

DISCUSSION

The AR_M values in Table IV can be used in order to describe the contribution of substituent groups to the lipophilicity of the whole molecule. As regards the Digitalis genins, the lipophilic character decreases in the order digitoxigenin > gitoxigenin > digoxigenin. In fact, attaching an OH group at C-12 (digoxigenin) or C-16 (gitoxigenin) decreases the lipophilic character of the parent compound, digitoxigenin. According to Dzimiri *et d.',* digoxigenin is more hydrophilic than gitoxigenin as the OH group at C-12 is more exposed to the complementary hydroxyl groups of the aqueous phase than the OH group at C-16. A similar conclusion can be drawn by considering the fragment values for the hydroxyl group at C-12 and C-16. In fact, the *Alog P* values for the pairs digoxigenin-digitoxigenin and gitoxigenin-digitoxigenin are -1.50 and -0.88 , respectively (Table V). Therefore, the $\Delta \log P$ value for the OH at C-12 is much closer to the aliphatic fragment value of -1.64 for the OH group¹⁸ than that for the OH at C-16. On the other hand, the acetyloxy group has the same hydrophilic character at both C-12 and C-16. Apparently the lactone ring is not able to mask the acetyloxy group at C-16. As a consequence, acetylation of the OH group has opposite effects at C-16 and C-12. k-Strophanthidin with an OH group at C-5 and a formyl group at C-10 is slightly more hydrophilic than digoxigenin. Ouabagenin (g-strophanthidin) is the most hydrophilic genin because of the addition of two other OH groups at C-l and C-l 1 and the replacement of the formyl group with a more hydrophilic $CH₂OH$ group at C-10. On the other hand, the replacement of the OH group at C-16 with a formyl group makes 16-formylgitoxin less lipophilic than gitoxin.

The sugar residues at C-3 examined in the present and Cohnen *et al's* system, *i.e.,* oleandrose, cymarose, digitoxose. rhamnose, thevetose and glucose, have polarities increasing in that order. As mentioned above, $Dayydot^6$ obtained the same order of ranking except for oleandrose and cymarose, for which the same retention value was obtained. The introduction of an α - and/or β -methyl group and an acetyl group(s) into the sugar side-chain increases the lipophilicity. It may be noted that our experimental R_M values for heptaacetyl-k-strophanthoside and pentaacetylgitoxin, 3.41 and 4.59, respectively, are very close to those calculated in the present system from the R_M value of k-strophantoside and 16-acetylgitoxin, 3.44 and 4.33, respectively. The use of the average AR_M for the acetyl group in the side-chain in the calculation of the *RM* value for strophantidin-3-acetate in the Cohnen *et al.* system may also be pointed out. This seems to be justified by the fact that in the present system the OCOCH₃ vs. OH group at C-3 has a AR_M value of 0.29 which is not far from the average AR_M of 0.35 reported in Table IV for any acetyl group in the side-chain.

The additivity of the lipophilic contribution of any substituent group at C-3 seems to rule out any significant kind of interaction between the steroid nucleus and the sugar side-chain. Fig. 1 shows histograms which illustrate the increments in lipophilicity due to the addition of the same sugar residue in each family of genins. It is shown that in each of the *Digitalis* derivatives families the lipophilicity increases in the

Fig. 1. Influence of side-chain composition on the overall lipophilicity of *Digigita1i.s* cardiac glycosides. The different subsets are (A) digitoxigenin, (B) gitoxigenin and (C) digoxigenin.

order genin \lt monodigitoxoside \lt bisdigitoxside \lt trisdigitoxoside $\lt \alpha$ - or β -acetyl $< \alpha$, β -diacetyl. Lanatoside A, B and C are less lipophilic than the α - or β -acetyl derivatives because of the presence of a glucosyl group in the side-chain.

A comment is deserved from a more general point of view. The present R_M values were obtained by extrapolation from the linear relationship between R_M values and acetone concentrations in the mobile phase. The extrapolation technique allows the calculation of a theoretical R_M value at 0% acetone in the mobile phase, *i.e.*, in a standard system where all the compounds could be compared.

Another great advantage over the determination of the R_M values at only one organic solvent concentration in the mobile phase is that in this way one can avoid the error that might arise because of different slopes of the straight lines describing the relationship between R_M values and organic solvent concentration in the mobile phase. Two compounds might have the same R_M value at a given organic solvent concentration and different extrapolated R_M values. Finally, the extrapolation procedure tends to yield a wider range of R_M values, which is reflected in the slopes of eqns. 3 and 4. In fact, the R_M values in the octanol system were measured at only one organic solvent concentration in the mobile phase.

CONCLUSION

The chromatographic data and the partition coefficients from the literature agree well with the present R_M values in describing the lipophilicity of cardiac steroids. The reliability of the AR_M values in Table IV is an important aspect. In fact, the fragment values can be used in calculating the R_M values of other cardiac steroids. Finally, this work seems to have demonstrated some advantages of reversed-phase TLC or any other chromatographic system over the use of the classical partition coefficient^{7,19}. The chromatographic method is simple and rapid; it requires little material, which is important with compounds that are expensive and/or difficult to synthesize, such as the cardiac steroids; the material does not need to be very pure; and the detection of spots by non-specific methods avoids the need for specific quantitative analytical methods.

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